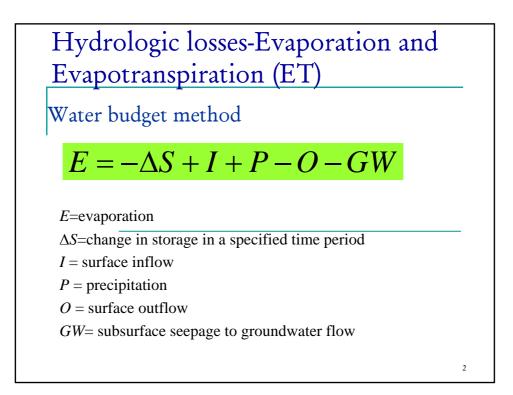
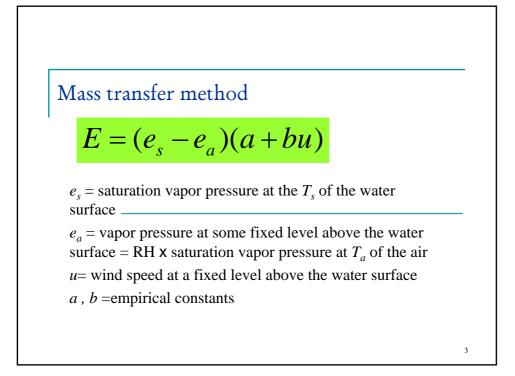
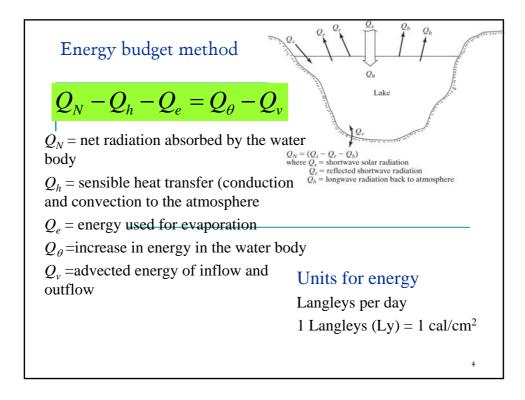
## Hydrologic Principles Part C

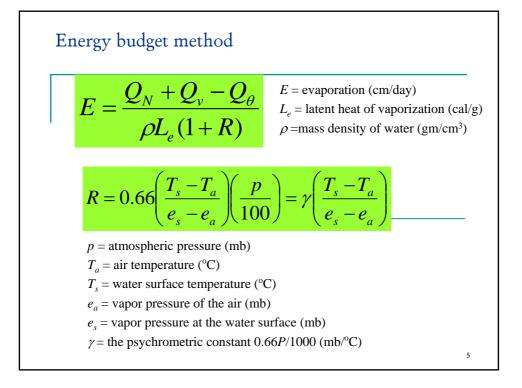
## Ahmad Sana

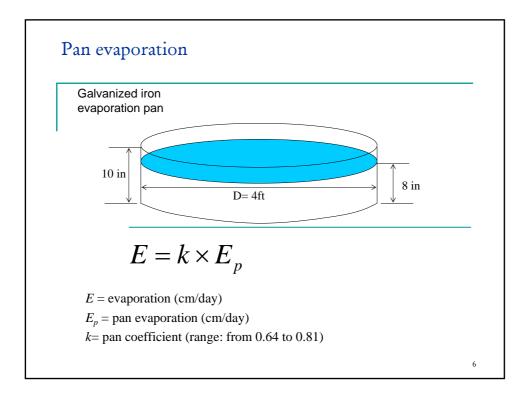
Department of Civil and Architectural Engineering Sultan Qaboos University Sultanate of Oman sana@squ.edu.om http://ahmadsana.tripod.com/



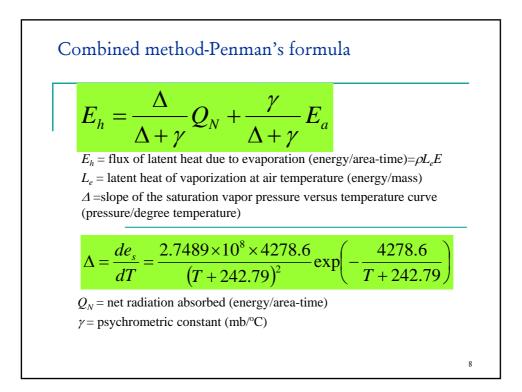


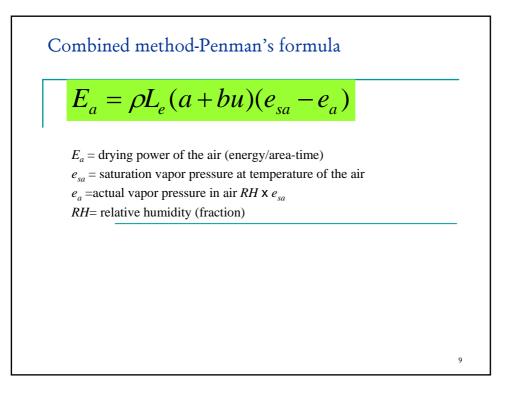




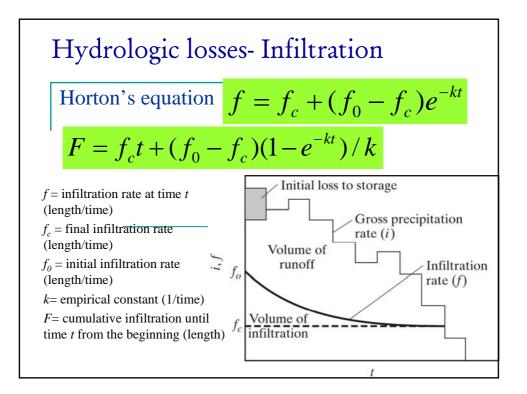


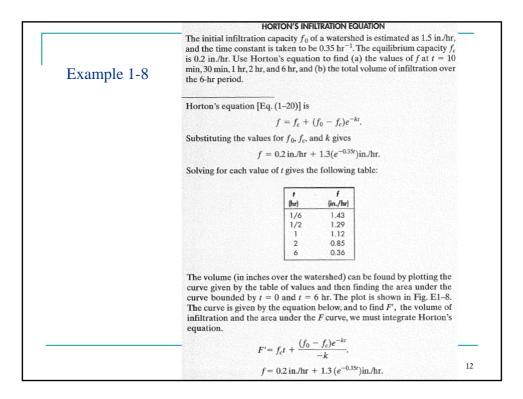
Type of Cover St. Augustine grass	0.77	Weaver and Stephens (1963
Bell peppers	0.85-1.04	weaver and Stephens (1905
Grass and clover	0.80	Brutsaert (1982, p. 253)
Oak-pine flatwoods (east Texas)	1.20	Englund (1977)
Well-watered grass turf		Shih et al. (1983)
Light wind, high relative humidity	0.85	
Strong wind, low relative humidity	0.35	
Everglades agricultural areas	0.65	
Irrigated grass pasture	0.76	Hargreaves and Samani
(central California)		(1982)

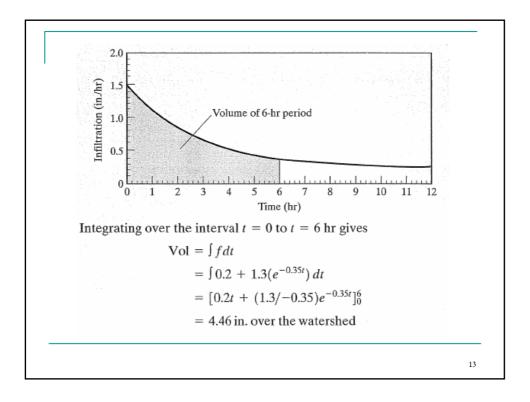


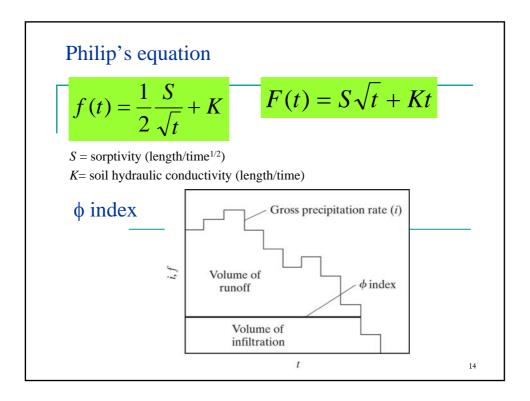


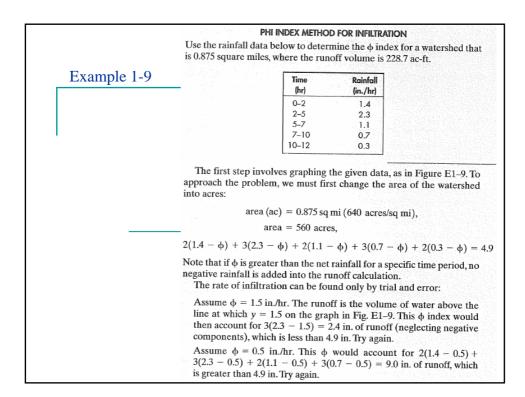
	EVAPORATION USING THE PENMAN EQUATION	
	An example of Eq. $(1-12)$ is given by Meyer (1944) for Minnesota lakes:	
	$E = 0.0106(1 + 0.1u)(e_s - e_a)$	
Example 1-7	with E in in./day, u in mph, and vapor pressures in mb. For an air tem- perature of 90°F (32.2°C), wind speed of 20 mph, relative humidity of 30%, and net radiation flux of 400 ly/day, estimate the evaporation rate using the Penman equation. Assume atmospheric pressure $\approx 1000$ mb, so the psychometric constant $\gamma = 0.66$ mb/°C.	
	We evaluate $\Delta$ from Eq. (1–18) at the air temperature of 32.2°C:	
	$\Delta = \frac{de_{\rm s}}{dT} = \frac{2.7489 \times 10^8 \times 4278.6}{\left(32.2 + 242.79\right)^2} \exp\left(-\frac{4278.6}{32.2 + 242.79}\right)$	
	$= 2.72 \text{ mb/}^{\circ}\text{C}.$	
	For Eq. $(1-17)$ we need the saturation vapor pressure at 32.2°C. From Eq. $(1-6)$ , we then have	
	$e_{sa} = 2.7489 \times 10^8 \exp[-4278.6/(32.2 + 242.79)] = 48.1 \text{ mb.}$	
	Thus, $e_a = e_{aa} = 0.3(48.1) = 14.4$ mb. The latent heat of vaporization at the air temperature of $32.2^{\circ}$ C is	
	$L_e = 597.3 - 0.57(32.2) = 579$ cal/g.	
	The density of water will be taken as $1 \text{ g/cm}^3$ . We include a change in units from in/day to cm/day while evaluating $E_a$ ,	
	$E_a = 0.0106(1 + 0.1 \cdot 20)(\text{in./day-mb})2.54(\text{cm/in.})1(\text{g/cm}^3)$	
	$\times$ 579(cal/g)(48.1 - 14.4)(mb)	
	$= 1590 \text{cal/cm}^2 \text{-day} = 1590 \text{ ly/day}.$	
	The Penman evaporation energy flux is thus	
	$E_h = 2.72/(2.72 + 0.66) \cdot 400 + 0.66/(2.72 + 0.66) \cdot 1590$	
	= 632 ly/day.	
	This can be converted to depth/day by dividing by $\rho L_e$ :	1
	E = 632/1(579) = 1.09 cm/day = 0.43 in./day of evaporation.	10

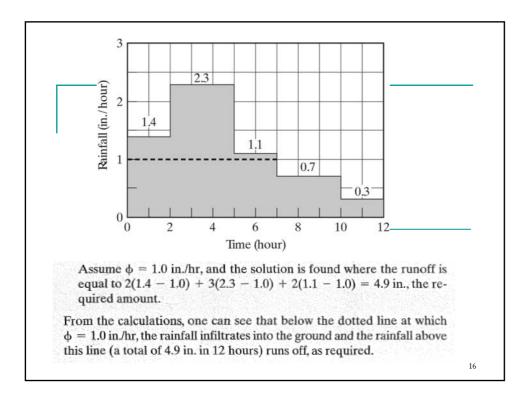


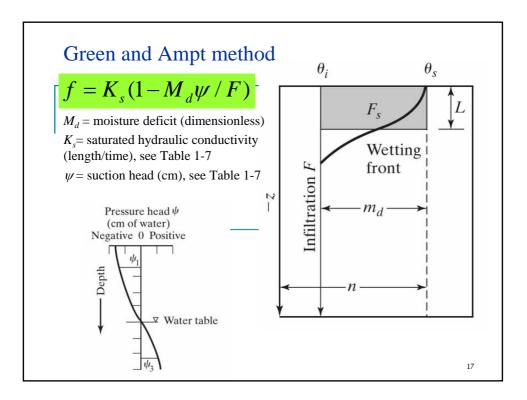


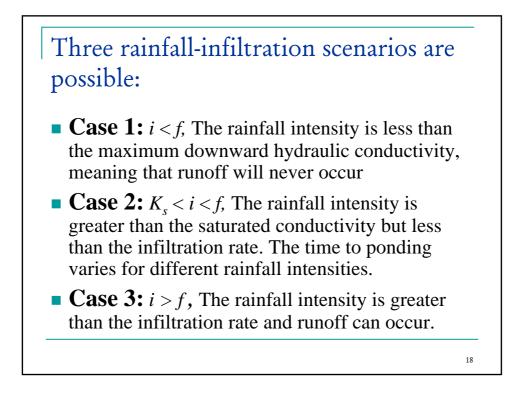


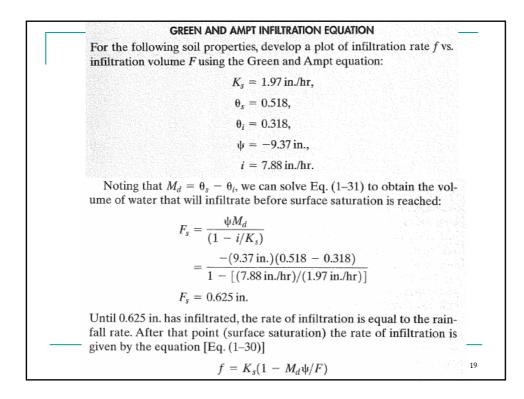


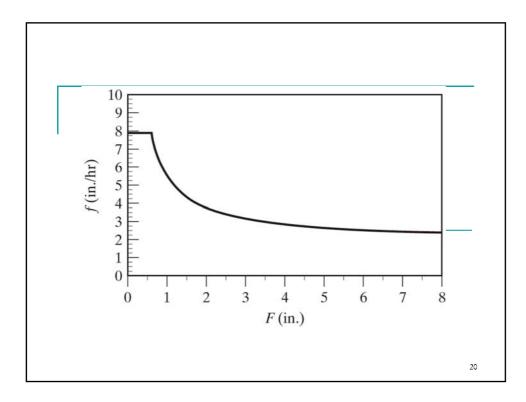


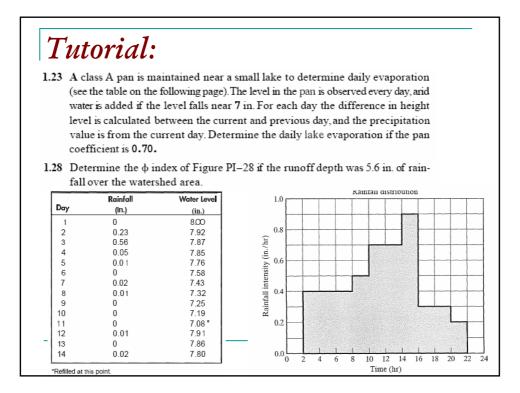












##